TRIMETHYLSILYLDIAZOMETHANE : A CONVENIENT REAGENT FOR THE O-METHYLATION OF ALCOHOLS¹

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Abstract: Trimethylsilyldiazomethane smoothly reacts with alcohols in dichloromethane in the presence of 42% aqueous fluoroboric acid to give methyl ethers in good to high yields.

Although a number of the methods for the O-methylation of alcohols are already known,² a novel general method is still required since the known methods have some drawbacks and limitations. We have already demonstrated that trimethylsilyldiazomethane (TMSCHN₂) is quite useful as a reagent for the methyl esterification of carboxylic acids³ and the O-methylation of phenols and enols.⁴ As an extension of these works, we now wish to report that TMSCHN₂ can be also used for the O-methylation of alcohols.

We have found that TMSCHN2 smoothly reacts with alcohols 1 in dichloromethane in the presence of 42% aqueous fluoroboric acid (FBA) to give the corresponding methyl ethers 2.

A typical experimental procedure is as follows: To a vigorously stirred mixture of 1 (1mmol) and FBA (209mg, 1mmol) in dichloromethane (4ml) was added dropwise TMSCHN2⁵ (1.8M hexane solution, 0.56ml, 1mmol) at 0°C during 4 min. The yellow color of TMSCHN2 immediately disappeared in contact with the reaction mixture with evolution of nitrogen. The stirring was continued at 0°C, and three further portions of TMSCHN2 (0.28ml (0.5mmol), 0.14ml (0.25mmol), and 0.14ml (0.25mmol)) were added dropwise at intervals of 20 min. The mixture was stirred at 0°C for further 30 min, poured into water, and extracted with dichloromethane. The organic layer was washed with water, dried over magnesium sulfate, and concentrated. The residue was purified by column chromatography on silica gel (BW-820MH, Fuji Davison) to give 2.

The results are summarized in Table. Various alcohols including primary, secondary, and tertiary ones smoothly react with TMSCHN2 to give 2. 2-Hydroxymethylcyclohexanone and 10-undecene-1-ol give the corresponding methyl ethers without any change of their ketone and carbon-carbon double bond functions, respectively. Monosaccharide such as methyl 2,3,6-tri-O-benzoyl-α-D-glucopyranoside also undergoes the O-methylation though sluggishly. Interestingly, in the case of N-benzylethanolamine, the selective N-methylation proceeds to give N-benzyl-N-methylethanolamine in 38% yield. Dichloromethane seems to be the reaction solvent of choice. The use of diethyl ether, benzene, or acetonitrile as the solvent gives poorer results.

Although 2-(1-naphthyl)ethanol reacts with TMSCHN2 in the presence of one equivalent of FBA (contained 6.7 equivalents of water) to give the methyl ether in 93% yield, the use of 0.05 equivalents of

FBA (contained 0.33 equivalents of water) gives a mixture of the methyl ether (44%) and 2-(1-naphthyl)ethyl trimethylsilyl ether (32%). These results suggest the reaction mechanism as follows:

Diazomethane is known to react with alcohols in the presence of catalytic amounts of concentrated FBA to give the methyl ethers.⁶ However, this method is very tedious and all manipulations should be carried out with great care. The present method using commercially available TMSCHN₂ is easy to conduct, appears to be general, and will provide a convenient method for the O-methylation of alcohols.

Tablea) O-Methylation of Alcohols with TMSCHN2

| Run | Starting Alcohol | Yield of product (%) | bp(°C)/mmHg ^{b)} or mp(°C) |
|-----|--|----------------------|--|
| 1 | 2-(1-Naphthyl)ethanol | 93 | 165-170/17 |
| 2 | 1-Dodecanol | 92 | 120-125/12 |
| 3 | 4-n-Butoxybenzylalcohol | 76C) | 115-125/3 |
| 4 | 1-Methyl-1-dodecanol | 84 | 90-95/1.5 |
| 5 | Dihydrocholesterol | 85 | 82-83d) |
| 6 | 1,1-Dimethyl-1-decanol | 74 | 75-80/0.9 |
| 7 | 10-Undecene-1-ol | 88 | 100-105/12 |
| 8 | Geraniol | 67 | 105-110/30 |
| 9 | Ethyl 12-Hydroxydodecanoate | 90 | 125-130/4 |
| 10 | 2-Hydroxymethylcyclohexanone | 84 | 110-115/26 |
| 11 | Methyl 2,3,6-Tri-O-benzoyl- α-D-glucopyranoside | 32(93)8) | |
| 12 | N-Benzylethanolamine | _f, g) | |

a) All products gave satisfactory spectral data and elemental analysis. b) By Kugelrohr distillation. c) Bis(4-n-butoxybenzyl) ether was also obtained in 12% yield. d) Lit., 6 mp 84-85°C. e) Yield in parentheses is based on recovered starting alcohol. f) Two equivalents of FBA was used, and the reaction was carried out in the presence of sodium chloride (50mg) for the purpose of salting-out. g) N-Benzyl-N-methylethanolamine was obtained as a sole isolable product in 38% yield.

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